

# Subunit Rotation in $F_0F_1$ -ATP Synthases as a Means of Coupling Proton Transport Through $F_0$ to the Binding Changes in $F_1$

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The rotation of an asymmetric core of subunits in  $F_0F_1$ -ATP synthases has been proposed as a means of coupling the exergonic transport of protons through  $F_0$  to the endergonic conformational changes in  $F_1$  required for substrate binding and product release. Here we review earlier evidence both for and against subunit rotation and then discuss our most recent studies using reversible intersubunit disulfide cross-links to test for rotation. We conclude that the  $\gamma$  subunit of  $F_1$  rotates relative to the surrounding catalytic subunits during catalytic turnover by both soluble  $F_1$  and membrane-bound  $F_0F_1$ . Furthermore, the inhibition of this rotation by the modification of  $F_0$  with DCCD suggests that rotation in  $F_1$  is obligatorily coupled to rotation in  $F_0$  as an integral part of the coupling mechanism.

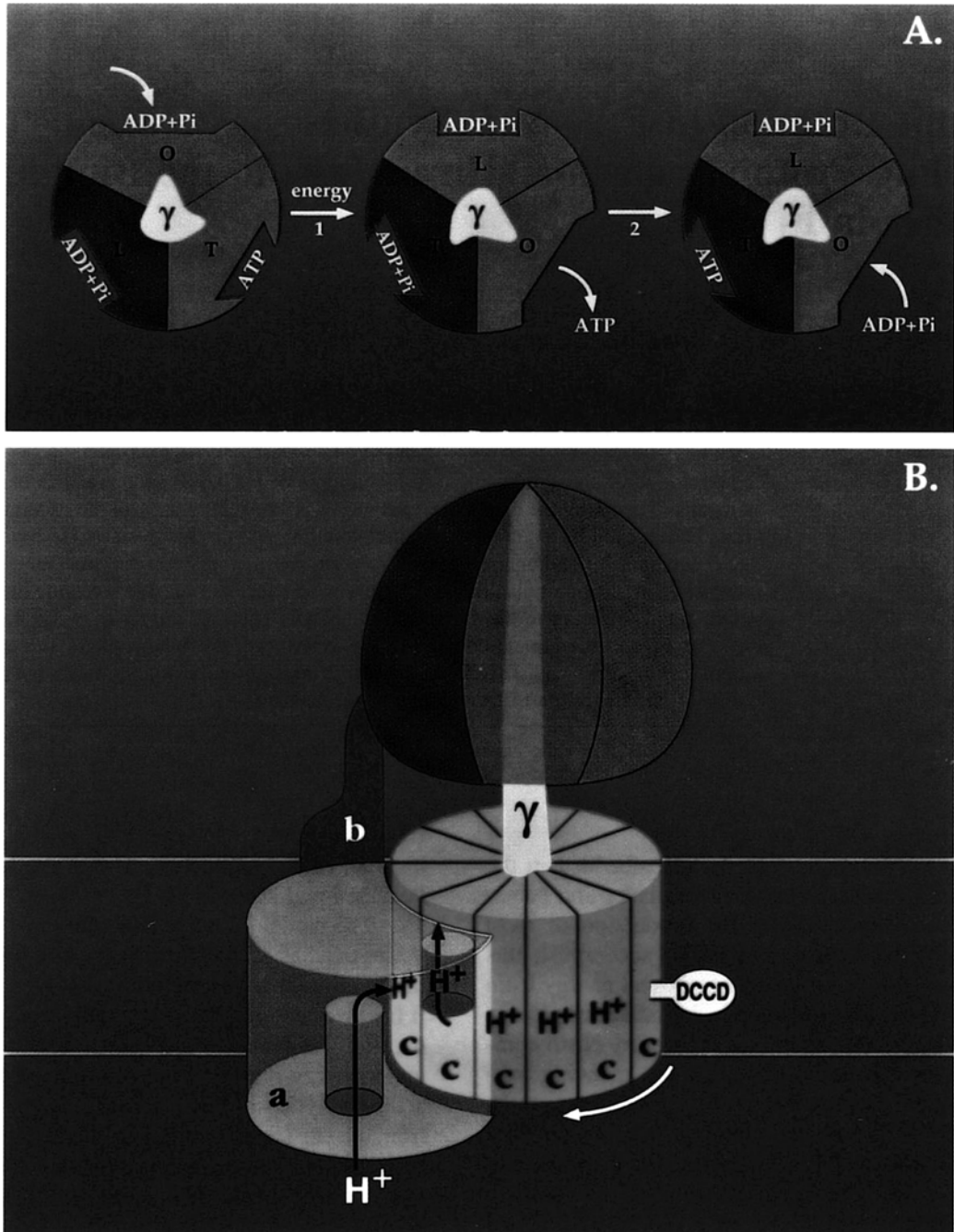
**KEY WORDS:** Binding change mechanism; subunit rotation; oxidative phosphorylation;  $F_0F_1$ -ATP synthase; rotary mechanism.

$F_0F_1$ -ATP synthases are found embedded in the membranes of mitochondria, chloroplasts, and bacteria, and are responsible for the production of most cellular ATP from ADP and  $P_i$ . To accomplish this critical task,  $F_0F_1$  must be able to extract energy from a transmembrane, electrochemical gradient of protons produced by photosynthetic or respiratory chain complexes.  $F_0$  is composed of membrane-spanning subunits and catalyzes the transport of protons across the bilayer.  $F_1$  is an extrinsic complex that contains the catalytic sites for ATP synthesis and hydrolysis, and can be removed from the membrane in a water-soluble form that catalyzes hydrolysis of ATP. The mechanism by which  $F_0F_1$  is able to achieve efficient, reversible energy coupling between a vectorial gradient of protons and the chemical energy of the terminal phosphoryl bond of ATP has been a major focus of research in bioenergetics for many years (for reviews see Fill-

ingame, 1990; Senior, 1990; Cross, 1992; Boyer, 1993; Pedersen and Amzel, 1993).

The model for energy coupling by  $F_0F_1$ -ATP synthases that has gained the most general support is called the binding change mechanism (Boyer, 1993, 1989). It has two features that are widely accepted (Boyer *et al.*, 1973; Kayalar *et al.*, 1977). The first is that the major energy-requiring step is not the synthesis of ATP at the catalytic site (Fig. 1a, step 2) but rather its release from the site (Fig. 1a, step 1). Second, the tight binding of substrates and release of product occur simultaneously at separate but interacting sites (Fig. 1a, step 1). A third premise has remained speculative (Boyer and Kohlbrenner, 1981). That is that the required binding changes are coupled to proton transport by the rotation of a complex of subunits that extends through  $F_0F_1$  (Fig. 1b).  $F_1$  has the subunit composition  $\alpha_3\beta_3\gamma\delta\epsilon$ , in which a hexamer of alternating  $\alpha$  and  $\beta$  subunits surrounds the central  $\gamma$  subunit. The catalytic nucleotide sites are located on the  $\beta$  subunits at  $\alpha/\beta$ -subunit interfaces. Rotation of the  $\gamma$  subunit in the center of  $F_1$  is thought to deform the surrounding catalytic subunits to give the binding changes (Fig. 1a). In  $F_0$ , with

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**Fig. 1.** A model for the binding change mechanism (adapted from Duncan *et al.*, 1995a). Panel a: View from the top of F<sub>1</sub>. In step 1, rotation of the asymmetric  $\gamma$  subunit forces conformational changes in the three catalytic subunits which result in affinity changes for substrates and product at the catalytic sites. T, L, and O stand for tight, loose, and open conformations and refer to decreasing affinities of catalytic sites for ligand. Each of the three blue/green areas represents a pair of  $\alpha$  and  $\beta$  subunits, where the catalytic sites are interfacial but mostly on  $\beta$ . In this illustration, the  $\alpha/\beta$  pairs remain stationary. In step 2, ATP forms spontaneously from tightly-bound ADP and P<sub>i</sub>. Panel b: View from the side of F<sub>0</sub>F<sub>1</sub>. Subunit *a* contains two partial channels, each in contact with a different side of the membrane. In order for a H<sup>+</sup> to traverse the membrane, it enters one channel, moves to the center of the membrane, transfers to the DCCD-reactive carboxyl on one of the *c* subunits, and is then carried to the other partial channel by rotation of the complex of *c* subunits. The *c* subunits are anchored directly to  $\gamma$ , whereas subunit *a* is anchored indirectly through subunit *b* to the catalytic subunits. Hence, the rotation of *c* subunits relative to subunit *a* in F<sub>0</sub> will drive the rotation of  $\gamma$  relative to the catalytic subunits in F<sub>1</sub>. The modification of a single *c* subunit per F<sub>0</sub> by DCCD, as shown, is sufficient to inhibit energy coupling.

subunit stoichiometry  $ab_2c_{9-12}$ , residues of both  $a$  and  $c$  subunits are thought to participate in proton transport, and rotation of the  $c$  subunits relative to the  $a$  subunit is believed to be required for completion of the proton pathway (Fig. 1b).

The idea of subunit rotation as a means of coupling the exergonic transport of protons through  $F_0$  to the endergonic conformational changes required for net synthesis of ATP by  $F_1$  arose as the result of two technical advancements. For the first 20 years of its application, the use of  $^{18}\text{O}$ -exchange methodology to follow the path of oxygen during oxidative- and photo-phosphorylation involved conversion of the oxygens of  $\text{P}_i$  to  $\text{CO}_2$ . Mass spectral analysis of the  $^{18}\text{O}$  content of the  $\text{CO}_2$  gave the average enrichment of  $^{18}\text{O}$  in the original  $\text{P}_i$  pool but provided no information regarding the  $^{18}\text{O}$  content of individual molecules of  $\text{P}_i$ . However, by the late 1970's, methods for esterifying the oxygens of  $\text{P}_i$  allowed  $\text{P}_i$  to be volatilized and analyzed directly. The ability to measure the distribution of individual species of  $\text{P}_i$  having zero to four  $^{18}\text{O}$  atoms provides a unique signature of the reaction pathway, one that reflects both the rate of reversible cleavage of ATP at the catalytic site and the rates of substrate binding and product release.

Since it was known that  $F_1$  contained multiple copies of the catalytic subunit but only single copies of the three smallest subunits,  $\gamma\delta\epsilon$ , it was assumed that the small subunits must have unique interactions with the catalytic subunits. It seemed reasonable to expect that this intrinsic asymmetry might give rise to sufficient differences in the rate constants at each of the three catalytic sites to allow detection of multiple catalytic pathways by  $^{18}\text{O}$ -exchange measurements. However, attempts to detect multiple pathways for enzyme from different sources failed (Hutton and Boyer, 1979; Hackney *et al.*, 1979; Kohlbrenner and Boyer, 1983).

One suggested solution to this apparent paradox was that the small subunits might possess tripartite structure so that their interactions with each catalytic subunit would be similar. This notion was dispelled by a second technical advancement. In 1981, sequences deduced from DNA were published for the  $\text{EcF}_1$  subunits (Saraste *et al.*, 1981; Kanazawa *et al.*, 1981; Nielsen *et al.*, 1981). The small subunits did not show repeat sequences.

These results led Paul Boyer to propose a second solution, namely that the asymmetric core of  $F_1$  rotates relative to the surrounding catalytic subunits (Fig. 1a, step 1). Hence in one complete cycle, each catalytic

subunit would have identical interactions with the asymmetric core, except that these interactions would be  $120^\circ$  out of phase (Boyer and Kohlbrenner, 1981; Boyer, 1983). However, a number of investigators in the field did not find the data compelling. It was argued that the asymmetric core of  $F_1$  might not cause a sufficient perturbation of the rate constants at the individual sites to allow detection of multiple catalytic pathways. In other words, structural asymmetry does not exclude the possibility of functional symmetry (Kironde and Cross, 1987; Musier and Hammes, 1987; Boyer, 1987).

As a test for rotation, the effects of cross-linking  $F_1$  subunits on ATP hydrolysis activity were measured. The results were mixed. Musier and Hammes (1987) reported that cross-linking the  $\gamma$  subunit of  $\text{CF}_1$  to an  $\alpha$  or  $\beta$  subunit did not correlate to a proportional loss in enzymatic activity. In light of more recent studies with  $\text{EcF}_1$  (Duncan *et al.*, 1995b; Aggeler *et al.*, 1995), these results are difficult to explain except to suggest that with soluble  $\text{CF}_1$ , rotation may be uncoupled from ATP hydrolysis (Cross, 1992). Using  $\text{EcF}_1$ , Kandpal and Boyer (1987) found that cross-linking  $\alpha$  to the small subunits did inhibit activity. Although these results are consistent with rotation, they do not prove it since intersubunit cross-links may simply prevent required conformational changes. The most clear cut experiment at this time was reported by Tozer and Dunn (1986). They found that an intersubunit disulfide bridge formed spontaneously between  $\delta$  and one of the  $\alpha$  subunits upon passage of  $\text{EcF}_1$  through a centrifuge column. No activity was lost. At the time, structural models of  $F_1$  placed all three of the single-copied subunits in the center of  $F_1$ , and the authors concluded that the asymmetric core could not rotate relative to the surrounding large subunits. However, we now know from a high-resolution structure (Abrahams *et al.*, 1994) that  $\gamma$  alone fills the center of the molecule. Therefore,  $\delta$  may be stationary relative to the large subunits.

Two additional approaches to test for subunit rotation were tried that were innovative in their design but flawed in theory. Moradi-Ameli and Godinot (1988) attached antibodies to the  $\alpha$  subunit of membrane-bound  $\text{MF}_1$ , increasing the mass of the  $F_1$  sector about twofold. There was no change in the rate of ATP synthesis, and the authors concluded that subunit rotation was unlikely. However, considering how rapidly protein molecules move at ambient temperature, rotation due to catalytic turnover on a millisecond time scale was far too slow to be affected by the increased mass. Musier-Forsyth and Hammes (1990) measured

the rotation of  $CF_0F_1$  in membranes using the time-resolved phosphorescence emission anisotropy of a covalently incorporated fluorescent probe. At 4°C the correlation time was 100–180  $\mu$ s. The value obtained under conditions for ATP synthesis was within the same range, and it was concluded that rotation has no role in catalysis. However, any rotation due to catalysis (on a sec time scale under the conditions used in this study) would have been orders of magnitude slower than the rate of rotation of  $CF_0F_1$  in the membrane.

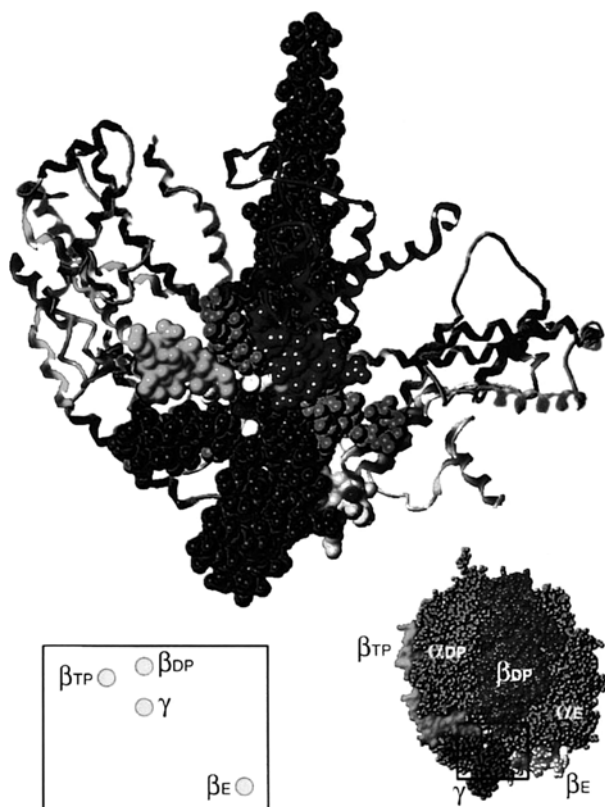
Evidence implicating a rotary-type mechanism also came from studies of  $F_0$ . Subunit *c* had long been known to contain a DCCD-reactive carboxyl group essential for proton transport (Hoppe and Sebald, 1984). This residue is predicted to be in the middle of a transmembrane  $\alpha$ -helix (Fig. 1b). In an attempt to define the path of the proton through the membrane, several laboratories mutated polar residues of the other two  $EcF_0$  subunits, *a* and *b*, predicted to be within the membrane phase. Several residues of subunit *a* proved essential (Cox *et al.*, 1986; Cain and Simoni, 1986; Lightowers *et al.*, 1988; Paule and Fillingame, 1989; Vik and Antonio, 1994), and the view emerged that the *a* and *c* subunits together form a proton channel. However,  $EcF_0$  has only one copy of subunit *a* but 9–12 copies of subunit *c* (Foster and Fillingame, 1982). All of the *c* subunits are required since modifying just one inhibits transport (Hermolin and Fillingame, 1989). This would appear to present a problem. How does a single *a* subunit coordinate with 9–12 *c* subunits to transport protons? One potential solution is that the *a* and *c* subunits rotate relative to each other (Cox *et al.*, 1986; Vik and Antonio, 1994; Hatch *et al.*, 1995). As each *c* subunit moves past the *a* subunit, a proton is transported (see caption to Fig. 1b).

The most supportive evidence for subunit rotation has come from studies of  $F_1$  structure. Cryoelectron microscopy by Capaldi and coworkers (Gogol *et al.*, 1990) showed that the central mass in  $EcF_1$  had an altered pattern of interactions with individual  $\beta$  subunits following catalytic turnover. The central mass is now known to be  $\gamma$ , and the results are consistent with its rotation. However, the resolution of the method used in this study was not sufficient to rule out the possibility that  $\gamma$  had simply wobbled rather than rotated. In contrast, the 2.8-Å resolution crystal structure for bovine  $MF_1$  (Abrahams *et al.*, 1994) shows  $\gamma$  to have unique interactions with each of the 3 catalytic subunits. In fact, it appears that the nature of the contact determines the conformational state of each catalytic

site as predicted by the third premise of the binding change mechanism (Boyer and Kohlbrenner, 1981).

In addition to providing supportive evidence for subunit rotation, the high-resolution structure also allowed a critical test of the concept. The structure identifies points of contact between  $\gamma$  and the  $\beta$  subunits. One of these includes the positioning of the bovine homolog of *E. coli*  $\gamma$ -subunit C87 close to one of the  $\beta$ -subunit  ${}_{380}DELSEED_{386}$  sequences. We mutated several residues in the DELSEED sequence, one at a time to Cys. The rationale was that the ability to form a reversible disulfide cross-link between  $\gamma$  and  $\beta$  could prove useful in assessing the role of subunit rotation in catalysis. Of the residues tested with both soluble and membrane-bound  $F_1$ ,  $\beta D380C$  reacted most readily with  $\gamma C87$  to form an intersubunit disulfide bridge in the presence of an oxidant (Duncan *et al.*, 1995a). Based on the high-resolution structure of  $MF_1$ , the relative orientation of  $\gamma C87$  and the three  $\beta D380C$ 's is shown in Fig. 2 (yellow atoms). Designations for the  $\beta$  subunits refer to different conformational states (Abrahams *et al.*, 1994) where the catalytic site is empty (E) or occupied by either an ADP (DP) or a nucleoside triphosphate (TP). The model shows that the Cys of  $\beta_E$  is far too distant (28 Å, sulfur–sulfur) to react with  $\gamma C87$  and, although the Cys on  $\beta_{TP}$  is only  $\sim 10$  Å from  $\gamma C87$ , it points away from  $\gamma C87$  and is sterically blocked by intervening atoms of  $\gamma$ . This suggests that the disulfide forms between  $\gamma C87$  and the Cys on  $\beta_{DP}$ , which are  $\sim 10$  Å apart through unobstructed space.

With soluble  $\beta D380C-F_1$ , we demonstrated a direct correlation between the cross-linking of  $\gamma$  to  $\beta$  and a loss of ATP hydrolysis activity. These results are consistent with, but not proof of, a need for  $\gamma$  to rotate in the center of  $F_1$ . As a direct test for rotation, we first cross-linked  $\gamma$  to one of the  $\beta$  subunits in  $\beta D380C-F_1$  and then, using a subunit dissociation/reassembly approach, we exchanged the two noncross-linked  $\beta$  subunits for radiolabeled  $\beta$ 's (Duncan *et al.*, 1995b). The  $\beta$ - $\gamma$  disulfide was then reduced and the enzyme incubated under different conditions before reoxidation. The question we asked was whether  $\gamma C87$  re-formed a cross-link to its original, nonradioactive partner or, if  $\gamma$  rotated during the incubation, could it now cross-link to radiolabeled  $\beta$ . In the absence of catalytic turnover, the majority of  $F_1$  molecules re-formed a disulfide between  $\gamma$  and the nonradioactive  $\beta$  subunit. However, following a brief episode of catalytic turnover, the level of radiolabel in the cross-linked product was consistent with the ability of  $\gamma C87$  to



**Fig. 2.** Putative orientation of  $\beta$ D380C residues relative to  $\gamma$ C87 in  $EcF_1$ . The molecular model is based on the crystal structure of bovine  $F_1$  (Abrahams *et al.*, 1994), but *E. coli* residue numbers are used here. The large representation shows the following residues in space-filling mode: all defined residues of  $\gamma$  (green),  $\beta_{376}$ LGMD $\Delta$ ELSEEDK $_{387}$  for  $\beta_{TP}$  (red), and  $\beta_{DP}$  (purple),  $\beta_{380}$ DELSEEDK $_{387}$  for  $\beta_E$  (orange), and the bovine residues analogous to  $\beta_{379}$ MDELSEEDK $_{387}$  for each  $\alpha$  subunit (blue). The above sequences for  $\gamma$ ,  $\beta_{TP}$ ,  $\beta_{DP}$ , and  $\alpha_E$  include all residues of the known  $F_1$  structure within 15 Å of  $\gamma$ C87's sulfur. For each  $\alpha$  and  $\beta$  subunit, the remainder of the C-terminal helical-bundle domain is represented by ribbons ( $\alpha$ -helices) and tubes. The sidechain of  $\beta$ D380 was changed to cysteine for each  $\beta$  in the model structure, and these three residues plus  $\gamma$ C87 are colored by atom type (C = gray, N = dark blue, S = yellow). The inset at lower right shows the complete space-filling model of  $F_1$ , to provide a frame of reference for the orientation of the molecule. Note that the small helix of  $\gamma$  (containing  $\gamma$ C87) extends to the left, and the "top" of  $F_1$  is tilted back through the plane of the page to make the desired cysteines visible. The subunits are labeled according to Abrahams *et al.* (1994). The box on the inset encloses the 4 cysteines. At lower left, a blow-up of this box shows the sulfur atoms in their proper orientation and identifies their subunit locations. Note that, in the model, the sulfur on  $\beta_{TP}$  is mostly obscured and that on  $\beta_E$  is partially obscured by the cysteine's  $C_\beta$  atom.

interact equivalently with each of the three  $\beta$  subunits. Since  $\gamma$ C87 cannot achieve equivalent interactions with each  $\beta$  subunit without rotation of the entire  $\gamma$

subunit within the core of  $F_1$  (see Fig. 2), the results provide compelling evidence for rotation.

To determine whether such rotation has a physiological role, it was important to extend these studies to membrane-bound  $F_0F_1$ . Hybrid  $F_1$  containing radiolabeled or epitope-tagged  $\beta$  subunits in the two noncross-linked positions was found to bind to  $F_1$ -depleted membranes and to restore coupled functions upon reduction of the disulfide bond. Again, an ATP-hydrolysis-dependent rotation of  $\gamma$  was observed. Furthermore, rotation of  $\gamma$  within  $F_1$  was blocked when  $F_0$  was modified by DCCD (Zhou *et al.*, 1996). This suggests that rotation in  $F_1$  is obligatorily coupled to rotation in  $F_0$  as an integral part of the coupling mechanism.

Many interesting details of rotational coupling need further clarification and confirmation. It will be of interest to determine whether rotation is strictly sequential, and whether the direction of rotation reverses when changing from ATP synthesis to ATP hydrolysis. It will be important to measure the rate of subunit rotation in order to confirm that it is fast enough to be an intermediate in the coupling process. Also, a direct demonstration of rotation in the  $F_0$  sector would provide additional compelling evidence for a rotary-type mechanism (Fig. 1b). Finally, a satisfactory understanding of the overall process will likely require a high-resolution structure for  $F_0$ .

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